PCT

WORLD INTELLECTUAL PROPERTY ORGANIZATION International Bureau



(51) International Patent Classification 7:	T	JNDER THE PATENT COOPERATION TREATY (PCT) (11) International Publication Number: WO 00/06660
C09J 151/00	A2	
3 •		(43) International Publication Date: 10 February 2000 (10.02.00)
(21) International Application Number: PCT/US (22) International Filing Date: 15 July 1999 (European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR,
(30) Priority Data: 09/126,130 30 July 1998 (30.07.98)	ī	Published Without international search report and to be republished upon receipt of that report.
(71) Applicant: THE DOW CHEMICAL COMPANY 2030 Dow Center, Abbott Road, Midland, MI 486		
(72) Inventor: MCGEE, Robert, Lee; 3606 Mary Jan Midland, MI 48642 (US).	ne Dri	ve,
(74) Agent: HOWARD, Dan, R.; The Dow Chemical (Patent Dept., P.O. Box 1967, Midland, MI 48 (US).	Compa 641-19	ny, 67
·		
		·
		DUESTIE EN LA WIENT HOROUGH DOLARITA FOR A DUESTING TO
POLYURETHANE FOAMS	SED A	DHESIVE FILM WITH IMPROVED POLARITY FOR ADHERING TO
(57) Abstract		
between a polyolefin grafted with an ethylenically unsatura and a polymeric and/or oligomeric polyamine. The reac	ated car	films that adhere well to polar materials contain the product of a reaction boxylic acid and/or an ethylenically unsaturated carboxylic acid anhydride oduct may be blended with an ethylene polymer, such as a low density o polar substrates, such as polyurethane foam. The adhesive compositions
	. •	
·		
+		
		·

FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AL	Albania	ES	Spain	LS	Lesotho	SI	Slovenia
AM	Armenia	FI	Finland	LT	Lithuania	SK	Slovakia
AΤ	Austria	FR	France	LU	Luxembourg	SN	Senegal
AU	Australia	GA	Gabon	LV	Latvia	SZ	Swaziland
AZ	Azerbaijan	GB	United Kingdom	MC	Monaco	TD	Chad
BA	Bosnia and Herzegovina	GE	Georgia	MD	Republic of Moldova	TG	Togo
BB	Barbados	GH	Ghana	MG	Madagascar	TJ	Tajikistan
BE	Belgium	GN	Guinca	MK	The former Yugoslav	, TM	Turkmenistan
BF	Burkina Faso	GR	Greece		Republic of Macedonia	TR	Turkey
BG	Bulgaria	HU	Hungary	ML	Mali	TT	Trinidad and Tobago
ВЈ	Benin	IE	Ireland	MN	Mongolia	UA	Ukraine
BR	Brazil	IL.	Israel	MR	Mauritania	UG	Uganda
BY	Belarus	IS	Iceland	MW	Malawi	US	United States of America
CA	Canada	lТ	Italy	MX	Mexico	UZ	Uzbekistan
CF	Central African Republic	JP	Japan	NE	Niger	VN	Viet Nam
CG	Congo	KE	Kenya	NL	Netherlands	YU	Yugoslavia
CH	Switzerland	KG	Kyrgyzstan	NO	Norway	zw	Zimbabwe
CI	Côte d'Ivoire	KP	Democratic People's	NZ	New Zealand		
CM	Cameroon		Republic of Korea	PL	Poland		
CN	China	KR	Republic of Korea	PT	Portugal		
CU	Cuba	KZ	Kazakstan	RO	Romania		
CZ	Czech Republic	LC	Saint Lucia	RU	Russian Federation		
DE	Germany	LI	Liechtenstein	SD	Sudan		-
DK	Denmark	LK	Sri Lanka	SE	Sweden		
EE	Estonia	LR	Liberia	SG	Singapore		

HIGH TEMPERATURE POLYOLEFIN BASED ADHESIVE FILM WITH IMPROVED POLARITY FOR ADHERING TO POLYURETHANE FOAMS

This invention relates to adhesive films, and more particularly to adhesive films that are formulated to adhere better to polar materials.

5

10

15

20

25

30

35

Ethylene acrylic acid (EAA) copolymers and maleic anhydride (MAH) modified polymers are known to adhere well with polar materials. However, adhesive films prepared with high amounts of EAA copolymers or MAH polymers tend to be too expensive for routine use in many applications, such as in the production of automotive parts like headliners. In an attempt to provide lower cost adhesive films that exhibit good adhesive properties with polar materials, others have blended polymers exhibiting good adhesion to polar materials with less expensive olefin polymers. However, these blends still tend to be too expensive for many applications, or to exhibit unacceptable adhesion to polar materials or both.

It is also known to incorporate a catalytic amount of an acidic agent or a basic agent into an adhesive composition that contains a grafted polyolefin in order to enhance adhesion. The catalytic agent is selected from monoalkyl phosphates, dialkyl phosphates, monoalkylamines, dialkylamines, trialkylamines, heterocyclic amino compounds, amino carboxylic acids and salts of such acids. Examples of amines include tributylamine and trioctylamine. A disadvantage in using these amines, both of relatively low molecular weight, as catalytic agents is that they tend to volatilize, especially at temperatures needed to melt blend the catalytic agents with EAA copolymers, MAH-modified polymers or both. Volatilization of the amine catalytic agents may present health, safety and environmental issues, all of which are desirably avoided.

It is also known in the art to react primary amino- or hydrazide-substituted ultraviolet (UV) absorbers with some or all of the anhydride groups of anhydride containing polymers or copolymers to form pendant stabilizer substituted imide or amic acid groups. An amic acid results from a reaction between an anhydride and an amine wherein the amine attacks one anhydride carbonyl group to produce an amide while leaving a second carbonyl group unreacted as a carboxylic acid. The resulting polymeric stabilizers may be used as is or as concentrates to stabilize other polymer systems. Polymeric or polymer bound stabilizers are not lost from a polymer system by volatilization, migration, or extraction, even at high temperatures. Although the polymeric stabilizers themselves are not subject to volatilization, they are prepared from a primary amino-substituted 2-hydroxybenzophenone, a primary amino-substituted salicylic acid ester or primary amino-substituted oxalic acid amide, all of which are low molecular weight molecules that could volatilize when melt blended with anhydride-containing polymers or copolymers. The polymeric stabilizers prepared by reacting primary amino compounds with anhydride containing polymers or copolymers are not known to be used as adhesives or adhesion promoters.

A first aspect of this invention is an adhesion promoter or adhesive that is a product of a reaction between a first reactant that is a polyolefin grafted with a graftable monomer that is an

ethylenically unsaturated carboxylic acid, an ethylenically unsaturated carboxylic acid anhydride or both and a second reactant that is a polymeric polyamine, an oligomeric polyamine or both.

A second, but related aspect of the invention is an adhesive composition that comprises from 50 percent to 80 percent by weight (weight percent) of an ethylene polymer that is substantially free of graft modification and from 20 to 50 weight percent of the adhesion promoter or adhesive of the first aspect. The adhesive composition has sufficient polarity to allow it to effectively bond to polar materials, such as polyurethane foam.

A third, also related aspect of the invention is an adhesive film prepared from the adhesive composition of the second aspect.

10

15

20

30

35

The invention also relates to methods for preparing adhesives, adhesive compositions and films by using the adhesion promoter of the first aspect or by reacting it with another component.

The invention further relates to a method of adhering a first substrate to a second substrate, the method comprising a first step of applying the adhesive composition of the first aspect to a surface of at least one of the substrates and a step of bringing at least one surface of each substrate together in such a manner as to dispose the adhesive composition at an interface between such surfaces.

The adhesive or adhesion promoter is the product of a reaction between two components. A first component is a polyolefin that is grafted with an ethylenically unsaturated carboxylic acid, an ethylenically unsaturated carboxylic acid anhydride or both. A second component is polymeric polyamine, an oligomeric polyamine or both. The grafted polyolefin may be formed by grafting an ethylenically unsaturated monomer onto a polyolefin using known methods such as that described in U.S. Patent No. 4,612,155.

Polyolefins suitable for use as base polymers for grafting include ethylene homopolymers and copolymers of ethylene and an alpha-olefin (α-olefin). A preferred grafted polyolefin, on account of its relatively low cost and commercial availability, as well as excellent performance in the adhesive compositions and films of this invention, is a MAH-modified linear low density polyethylene (MAH-g-LLDPE). A suitable, commercially available MAH-g-LLDPE is sold under the name BynelTM 4104 by E. I. du Pont de Nemours and Company. Other polyolefins that may be grafted include copolymers of ethylene and a vinyl alkanoate, such as ethylene/vinyl acetate (EVA) copolymers; ethylene/alkyl(meth)acrylate copolymers including ethylene/methacrylate copolymers, ethylene/methyl acrylate (EMA) copolymers, ethylene/butyl acrylate (EBA) copolymers and ethylene/methylmethacrylate (EMMA) copolymers); ethylene/carbon monoxide (ECO) copolymers; ethylene/vinyl alkanoate/carbon monoxide terpolymers, such as ethylene/vinyl acetate/carbon monoxide (EVACO) terpolymers; and ethylene/alkyl(meth)acrylate/carbon monoxide terpolymers, such as ethylene/butyl acrylate/carbon monoxide (EBACO) terpolymers. Such copolymers are known in the art and many are commercially available.

The monomer grafted onto the polyolefin is at least one ethylenically unsaturated monomer selected from ethylenically unsaturated carboxylic acids and ethylenically unsaturated

carboxylic acid anhydrides. Examples of such acids and anhydrides include mono-, di- or polycarboxylic acids such as acrylic acid, methacrylic acid, maleic acid, fumaric acid, itaconic acid, crotonic acid, itaconic anhydride, Nadic Methyl Anhydride™ (methyl-bicyclo[2.2.1]heptene-2,3- dicarboxylic anhydride isomers, allied chemical corporation), MAH and substituted MAH materials, such as dimethyl MAH.

Suitable polymeric and oligomeric polyamines include dimers, trimers and tetramers (known collectively as oligomers), and polymers (macromolecules comprised of five or more monomers). The polymeric and oligomeric polyamines include at least two amine groups per molecule. The polyamines used for preparing the adhesive films of this invention preferably contain from 20 to 100 weight percent amine-containing monomer, more preferably at least 50 weight percent amine-containing monomer, based on polyamine weight. The polymeric and oligomeric polyamines employed in the practice of this invention preferably have a weight average molecular weight (M_W) of at least 250 daltons, more preferably at least 500 daltons, and most preferably at least 1,000 daltons.

10

15

20

25

30

35

A suitable polymeric polyamine is a copolymer of N,N'-bis(2,2,6,6)-tetramethyl-4-piperidinyl-1,6-hexadiamine with 2,4-dichloro-6-(4-morpholinyl)-1,3,5-triazine (a hindered amine type UV stabilizer commercially available under the trade designation Cyasorb® UV-3346 from Cytec Industries). Any utility of such a polyamine as a UV stabilizer in compositions of the present invention is merely incidental, as it is primarily, if not solely, used to enhance adhesion of the grafted polyolefins, and polymer blends containing such grafted polyolefins.

Polymeric and oligomeric polyamines can be prepared, for example, by solution polymerization, in aqueous media, of at least one ethylenically unsaturated monomer that includes an amine group, either with or without an ethylenically unsaturated comonomer that does not contain an amine group. Monomers that contain an amine group include aminoalkyl vinyl ethers or sulfides wherein alkyl groups are straight-chains or branched-chains and the nitrogen atom is a primary, secondary, or tertiary nitrogen atom. Specific examples including beta (β)-aminoethyl vinyl ether, βaminoethyl vinyl sulfide, N-monomethyl- β-aminoethyl vinyl ether or sulfide, N-monoethyl- βaminoethyl vinyl ether or sulfide, N-monobutyl- β-aminoethyl vinyl ether or sulfide, and N-monoethyl-3aminopropyl vinyl ether or sulfide. Other suitable monomers containing an amine group include acrylamides and aminoalkyl(methyl)acrylates, such as dimethylaminoethylacrylate or methacrylate, βaminoethyl acrylate or methacrylate, N- \u03b3-aminoethyl acrylamide or methacrylamide. N-(monomethylaminoethyl)-acrylamide or methacrylamide, N-(mono-n-butyl)-4-aminobutyl acrylate or methacrylate, methacryloxyethoxyethylamine, and acryloxypropoxypropylamine. Monomers that contain an amine group also include N-acryloxyalkyl-oxazolidines and N-acryloxyalkyltetrahydro-1,3oxazines and corresponding components in which the alkyl linkage is replaced by an alkoxyalkyl or a poly(alkoxy-alkyl). Examples include oxazolidinylethyl methacrylate, oxazolidinylethyl acrylate, 3-(gamma-methacryl-oxypropyl)-tetrahydro-1,3-oxazine, 3-(β-methacryloxyethyl)-2,2-penta-methyleneoxazolidine, 3-(β-methacryloxyethyl-2-methyl-2-propyloxazolidine, N-2-(2-acryloxyethoxy)ethyloxazolidine, N-2-(2-methacryloxyethoxy)ethyloxazolidine, N-2-(2-methacryloxyethoxy)ethyl-5-methyl-

oxazolidine, N-2-(2-acryloxyethoxy)ethyl-5-methyl-oxazolidine, 3-[2-(2-methacryloxyethoxy) ethyl)]-2,2-penta-methylene-oxazolidine, 3-[2-(2-methacryloxyethoxy)ethyl)]-2,2-dimethyloxazolidine, 3-[2(methacryloxyethoxy)ethyl]-2-phenyl-oxazolidine. Monomers that readily generate amines by hydrolysis are also useful. Examples of such monomers include acryloxy-ketimines and acryloxy-aldimines, with illustrative compounds including 2-[4-(2,6-dimethylheptylidene)-amino]-ethyl methacrylate, 3-[2-(4-methylpentylidine)-amino]-propyl methacrylate, β-)benzylideneamino)-ethyl methacrylate, 3-[2-(4-methylpentylidene)-amino]-ethyl methacrylate, 2-[4-(2,6-dimethylheptylidene)-amino]-ethyl acrylate, 12-(cyclopentylidene-amino)-dodecyl methacrylate, N-(1,3-dimethylbutylidene)-2(2-methacryloxyethoxy)-ethylamine, N-(benzylidene)-methacryloxyethoxyethoxy)-ethylamine, N-(1,3-dimethylbutylidene)-2-(2-acryloxyethoxy)-ethylamine, N-(benzylidene)-2-(2-acryloxyethoxy)-ethylamine.

Polymeric and oligomeric polyamines can be prepared by polymerizing any of the ethylenically unsaturated amine-containing monomers set forth above, either by themselves, with each other, or with an ethylenically unsaturated comonomer that lacks an amine group. Examples of amine-free comonomers include alkyl acrylates (esters of acrylic acid with an alcohol containing 1 to 18 carbon atoms (C₁₋₁₈), especially C₁₋₄ alcohols or alkanols), styrene, vinyl toluene, vinyl acetate, vinyl chloride, vinylidene chloride, substituted styrenes, butadiene, substituted butadienes, ethylene, and the nitriles and amides of acrylic and methacrylic acid.

10

1.5

20

25

30

35

The grafted polyolefins used in preparing adhesives and adhesion promoters need not have a particular carboxylic acid or carboxylic acid anhydride content. The grafted polyolefin preferably contains a total amount of 0.001 percent to 5 weight percent, based on grafted polyolefin weight, of carboxylic acid groups, carboxylic acid anhydride groups or both. Amounts below 0.001 weight percent lack desired adhesive properties. Amounts greater than 5 weight percent provide no significant additional benefits and amounts substantially in excess of 5 weight percent may actually detract from adhesive properties.

The polymeric polyamines, oligomeric polyamines or both are present in an amount that depends upon both desired properties and amine group-containing monomer content of such polyamines. The amount and type of polyamine preferably provide an excess of ethylenically unsaturated carboxylic acid groups, ethylenically unsaturated carboxylic acid anhydride groups or both. Generally, the amount and type are selected to yield a ratio of carboxylic acid groups, carboxylic acid anhydride groups or both to amine groups of from 10:1 to 3:2, preferably from 3:1 to 2:1.

The adhesive or adhesion promoter has a weight ratio of grafted polyolefin to polyamine of from 20:1 to 400:1. The adhesive or adhesion promoter may be used alone or, more preferably, it may be combined with an ethylene polymer that has not been graft modified to form an adhesive composition suitable for use in making an adhesive film.

In a preferred aspect of this invention, adhesive compositions comprise polymeric blends of from 50 weight percent to 80 weight percent of an ethylene polymer that has not been graft modified, and from 20 weight percent to 50 weight percent of the adhesive or adhesion promoter

disclosed above, the amounts being based upon composition weight and totaling 100 weight percent. The adhesive compositions can be prepared by combining an ethylene polymer that has not been graft modified and the adhesion promoter in a conventional melt blending apparatus such as in an extruder. An alternate procedure combines the unmodified ethylene elastomer with those components that, when reacted, form the adhesive or adhesion promoter in such an apparatus. Irrespective of the procedure, the blends typically have a polyamine content of from 0.1 to 1 weight percent, based upon total blend weight.

"Ethylene polymer", as used herein, refers to ethylene homopolymers and copolymers of ethylene and at least one other ethylenically unsaturated monomer. Examples of ethylene copolymers include copolymers of ethylene and at least one higher (from 3 to 20 carbon atoms (C₃₋₂₀), preferably C₃₋₈)α-olefin monomer such as propylene, butene-1, hexene-1 and octene-1; copolymers of ethylene and a vinyl alkanoate such as EVA copolymers and/or copolymers of ethylene and alkyl (meth)acrylate. Preferred ethylene polymers, on account of their relatively low cost, commercial availability, and satisfactory performance in the adhesive compositions and films of this invention, are low density polyethylenes (LDPE). A particularly suitable, commercially available LDPE is LDPE 620l (The Dow Chemical Company). Other suitable ethylene polymers include those identified above as suitable base polymers for grafting.

10

15

20

25

30

35

The adhesive compositions of this invention may, and preferably do, contain a silicon dioxide (SiO₂) antiblock concentrate. A suitable SiO₂antiblock concentrate is CN 734 (Southwest Plastics, Houston Texas). The antiblock concentrate may be used in an amount up to 7.5 weight percent, based on adhesive composition weight. Other additives, such as colorants, antioxidants and UV stabilizers, may be added in minor amounts as desired.

The adhesive compositions described above may be cast or extruded through a die to form adhesive films that exhibit enhanced polarity as determined by contact angle measurements using a Tantec™ contact angle meter model CAM-FILM, and improved adhesion with polar substrates as determined by measuring peel strength of adhesive films bonded to polyurethane foam.

The adhesive compositions of this invention exhibit improved (relative to adhesive films made from resin blends of EAA and LLDPE) adhesion to polar substrates and can be advantageously used to adhere or bond a first substrate to a second substrate, especially when at least one of the substrates has a polar surface. The adhesive compositions of this invention are especially useful for adhering a polyurethane material to another material, which may or may not also be a polar material, and which may or may not be polyurethane. An illustration of the latter involves bonding a polyurethane foam layer to another layer of an automobile interior panel, such as an automobile headliner. The adhesive composition is preferably applied to a substrate surface as a preformed film.

The following example illustrates certain aspects of the invention and does not limit the scope of the invention described in the specification and claims.

EXAMPLE

10

15.

20

25

Prepare a dry blend of polymer resin pellets containing 70 weight percent LDPE (LDPE 620I, The Dow Chemical Company), 20 weight percent MAH-g-LLDPE (BYNEL® 4104, E. ¶. du Pont de Nemours and Company), 5 weight percent of an antiblock concentrate (CN-734, Southwest Plastics) that consists of 85 weight percent LDPE and 15 weight percent SiO₂, and 5 weight percent of an additive concentrate (CN-720, Southwest Plastics) that consists of 87 weight percent LDPE, 10 weight percent of an oligomeric hindered amine that sometimes serves as a UV stabilizer (CYRASORB® 3346), 2 weight percent of a phosphite antioxidant (MARK® 2112) and 1 weight percent of a hindered phenolic antioxidant (CYANOX® 1790). The dry blend percentages are based on blend weight and concentrate percentages on concentrate weight. In each instance, the percentages total 100 weight percent.

Extrude the dry blend into a round profile that has a diameter of 0.21 inches (in) (0.53 centimeters (cm)) using a 2 inch (5.1 cm) single screw extruder. Operate the extruder at a rate of 20 revolutions per minute (rpm) using the following temperature profile: Barrel Zone (BZ) 1 - 332° Fahrenheit (°F) (166.7° Centigrade (°C)), BZ2 - 346°F (174.4°C), BZ3 - 366°F (185.5°C), Zone 4 (gate) - 370°F (187.8°C), Gate/adapter - 371°F (188.3°C), and Transfer line - 371°F (188.3°C). This profile results in an extrudate melt temperature of 381°F (193.9°C) and a barrel pressure of 2690 pounds per square inch (psi) (18.5 megapascals (Mpa)).

Press the extrudate between polyethylene terephthalate (PET) (Mylar®, E. I. du Pont de Nemours and Company) films at a temperature of 340°F (171.1°C), using an applied pressure of 15,000 pounds (lbs) (6804 kilograms (kg)), for 2 minutes and then cool the extrudate while maintaining the applied pressure for 2 minutes. Press additional films as identified in the table shown below under the same conditions. Remove the pressed extrudate films from the PET films and test them for contact angle of water on a pressed extrudate film surface using a goniometer. See, S. Wu, Polymer Interface and Adhesion, Marcel Dekker, Inc., New York and Basel, (1992) pp. 178-181 for surface tension calculations and methodology.

Make contact angle measurements at room temperature using deionized water bubbles at bubble apex with a Tantec contact angle meter model CAM-FILM. The results shown in the following table represent an average of three measurements.

FILM	CONTACT ANGLE
620I LDPE	82.7° ± 4.2°
Bynel® 4104	60.0° ± 0°
CN-734	78.0° ± 2.0°
CN-720	50.0° ± 10.0°
Composition of the Invention	65.3° ± 1.2°

30

The above results show that an adhesive film prepared from an adhesive composition in accordance with this invention, containing a relatively high amount (70 weight percent) of non-wetable LDPE and relatively low amounts of polar materials (Bynel® 4104 and CN-720), is very polar.

By way of contrast, contact angle measurements of a film that lacks a polyamine (additive concentrate CN-720) average 82.7° ± 2.52°. Corona discharge treatment of one side of the film to 42 dynes/centimeter reduces the contact angle to 72.0° ± 0.0°. The film composition is 64 weight percent LDPE, 30 weight percent MAH-g-LLDPE and 6 weight percent antiblock concentrate (CN-734), all percentages based on composition weight and totaling 100 weight percent.

The foregoing results demonstrate that the presence of an oligomeric polyamine (contained in the additive concentrate CN-720) provides a significant reduction in contact angle. This equates to a significant increase in polarity relative to films that lack an oligomeric polyamine, a polymeric polyamine or both. The results show that inclusion of a polyamine provides a greater reduction in contact angle than that resulting from a corona discharge treatment.

Conduct adhesion tests by bonding rigid polyurethane foam to flexible polyurethane foam backed fabric with a pressed extrudate film prepared as detailed above using a Sentinel heat sealing bar set to 280°F (137.8°C) and an applied force of 30 psi (206.8 kilopascals (Kpa)) for 20 seconds. Cut test strips (1 inch or 2.5 cm wide) from the bonded foams and peel the foams apart using an Instron tensile testing machine. The following table contains numerical values that represent an average of values from five test samples.

Maximum Load

Average Load Between Limits

Failure Mode

2.13 lb ± 0.27 lb (0.97 ± 0.12 kg) 1.76 lb ± 0.23 lb (0.80 ± 0.10 kg) Cohesive

20

25

5

10

15

The above results demonstrate that the adhesive films of this invention exhibit excellent adhesion to polar substrates such as polyurethane foams, with the failure mode being cohesive, not adhesive.

It will be apparent to those skilled in the art that various modifications to the preferred embodiment of the invention as described herein can be made without departing from the spirit or scope of the invention as defined by the appended claims.

1. An adhesion promoter composition comprising a product of a reaction between a first reactant that is a polyolefin grafted with a graftable monomer that is an ethylenically unsaturated carboxylic acid, an ethylenically unsaturated carboxylic acid anhydride or both, and a second reactant that is a polymeric polyamine, an oligomeric polyamine or both.

2. The adhesion promoter of Claim 1, wherein the first reactant is a maleic anhydride modified linear low density polyethylene.

5

10

15

20

25

30

35

- 3. The adhesion promoter of Claim 1, wherein the polyolefin of the first reactant is selected from the group consisting of copolymers of ethylene and a vinyl alkanoate, ethylene/alkyl(meth)acrylate copolymers, ethylene/carbon monoxide copolymers, ethylene/vinyl alkanoate/carbon monoxide terpolymers, and ethylene/alkyl(meth)acrylate/carbon monoxide terpolymers.
- 4. The adhesion promoter of Claim 1, wherein the graftable monomer is selected from the group consisting of acrylic acid, methacrylic acid, maleic acid, fumaric acid, itaconic acid, crotonic acid, itaconic anhydride, nadic anhydride, maleic anhydride, and substituted maleic anhydrides.
- 5. The adhesion promoter of Claim 1, wherein the second reactant has an amine-containing monomer content of from 20 percent to 100 percent by weight, based on second reactant weight.
- 6. The adhesion promoter of Claim 1, wherein the polyamine has a molecular weight of at least 250 daltons.
- 7. The adhesion promoter of Claim 1, wherein the polyamine is a copolymer of N,N'-bis(2,2,6,6)-tetramethyl-4-piperidinyl-1,6-hexadiamine with 2,4-dichloro-6-(4-morpholinyl)-1,3,5-triazine.
- 8. The adhesion promoter of Claim 1, wherein the first reactant has a graftable monomer content of from 0.001 percent to 5 percent by weight, based on first reactant weight.
- 9. The adhesion promoter of Claim 1, wherein the first reactants are present in amounts that provide a ratio of graftable monomer to amine groups of from 10:1 to 3:2.
- 10. An adhesive composition comprising from 50 percent by weight to 80 percent by weight of an ethylene polymer that is substantially free of graft modification and from 20 percent by weight to 50 percent by weight of the adhesion promoter of any of Claims 1-9.
- 11. The adhesive composition of Claim 10, wherein the ethylene polymer that is substantially free of grafting is selected from the group consisting of a low density polyethylene (LDPE), a copolymer of ethylene and at least one higher alpha-olefin monomer, an ethylene carbon monoxide copolymer, an ethylene/vinyl alkanoate/carbon monoxide terpolymer.
- 12. The adhesive composition of Claim 10 further comprising an antiblock concentrate.
- 13. A method of adhering a surface of a first substrate to a surface of a second substrate, comprising:

applying the adhesive composition of Claim 10 to the surface of at least one of the first and second substrates e; and

bringing at least one surface of each substrate together in such a manner as to dispose the adhesive composition at an interface between such surfaces.

- 14. The method of Claim 13, wherein at least one of the substrates comprises a polar material.
 - 15. An adhesive film fabricated from the adhesive composition of Claim 10.

(19) World Intellectual Property Organization International Bureau



(43) International Publication Date 10 February 2000 (10.02.2000)

(10) International Publication Number WO 00/06660 A3

pany, Patent Dept., P.O. Box 1967, Midland, MI 48641-

(74) Agent: HOWARD, Dan, R.; The Dow Chemical Com-

(81) Designated States (national): AU, BR, CA, CN, JP, KR,

(84) Designated States (regional): European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC,

C09J 151/06

(21) International Application Number: PCT/US99/16048

(22) International Filing Date: 15 July 1999 (15.07.1999)

(25) Filing Language:

(51) International Patent Classification7:

English

(26) Publication Language:

English

(30) Priority Data: 09/126,130

30 July 1998 (30.07.1998)

US

C08G 81/02,

(71) Applicant: THE DOW CHEMICAL COMPANY [US/US]; 2030 Dow Center, Abbott Road, Midland, MI 48674 (US).

(72) Inventor: MCGEE, Robert, Lee; 3606 Mary Jane Drive, Midland, MI 48642 (US).

Published: with international search report

1967 (US).

MX. PL.

NL, PT, SE).

(88) Date of publication of the international search report: 13 December 2001

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(54) Title: HIGH TEMPERATURE POLYOLEFIN BASED ADHESIVE FILM WITH IMPROVED POLARITY FOR ADHER-ING TO POLYURETHANE FOAMS

(57) Abstract: Adhesion promoters, adhesive compositions, and adhesive films that adhere well to polar materials contain the product of a reaction between a polyolefin grafted with an ethylenically unsaturated carboxylic acid and/or an ethylenically unsaturated carboxylic acid anhydride, and a polymeric and/or oligomeric polyamine. The reaction product may be blended with an ethylene polymer, such as a low density polyethylene, to form an adhesive composition that bonds well to polar substrates, such as polyurethane foam. The adhesive compositions may be extruded or cast to form adhesive films.

INTERNATIONAL SEARCH REPORT

International Application No

		ru	1/03 99/10048
A. CLASSIF IPC 7	COSGS1/02 CO9J151/06		
According to	International Patent Classification (IPC) or to both national classifica	ation and IPC	
B. FIELDS	SEARCHED		
Minimum do	cumentation searched (classification system followed by classification COSG CO9J COSL B32B	on symbols)	
	on searched other than minimum documentation to the extent that si	·	
Electronic da	ata base consulted during the international search (name of data bas	se and, where practical, sear	ch terms used)
C. DOCUME	ENTS CONSIDERED TO BE RELEVANT		
Category *	Citation of document, with indication, where appropriate, of the rel	evant passages	Relevant to claim No.
Х	US 4 822 688 A (NOGUES PIERRE) 18 April 1989 (1989-04-18) claims 1,2		1,4,8,9
X	EP 0 177 401 A (ATOCHEM) 9 April 1986 (1986-04-09) claims 1,3,4		1,4,8,9
x	PATENT ABSTRACTS OF JAPAN vol. 018, no. 430 (C-1236), ll August 1994 (1994-08-11) & JP 06 128429 A (MITSUI PETROCHE LTD), 10 May 1994 (1994-05-10) abstract	M IND	1
А	EP 0 322 138 A (EXXON CHEMICAL PAINC) 28 June 1989 (1989-06-28) claims 1,3	ATENTS	1
Furt	her documents are listed in the continuation of box C.	X Patent family mem	bers are listed in annex.
Special c	ategories of cited documents:	T later document published	after the international filling date
consid	ent defining the general state of the art which is not dered to be of particular relevance document but published on or after the International	or priority date and not cited to understand the invention	in conflict with the application but principle or theory underlying the
filing of "L" docume which		cannot be considered in involve an inventive ste "Y" document of particular re	ovel or cannot be considered to p when the document is taken alone elevance; the claimed invention
"O" docum other	ent referring to an oral disclosure, use, exhibition or means	document is combined	o involve an inventive step when the with one or more other such docu- on being obvious to a person skilled
"P" docum later t	ent published prior to the international filing date but than the priority date claimed	"&" document member of the	e same patern family
Date of the	actual completion of the international search	Date of mailing of the In	ternational search report
1	3 October 1999	25/10/1999	
Name and	mailing address of the ISA European Patent Office, P.B. 5818 Patentlaan 2 NL – 2280 HV Rijswijk	Authorized officer	·
	NL – 2250 HV HISWIK Tel. (+31-70) 340–2040, Tx. 31 651 epo ni, Fax: (+31-70) 340–3016	Niaounakis	, M

INTERNATIONAL SEARCH REPORT

'nformation on patent family members

International Application No Pt./US 99/16048

	tent document in search report		Publication date		atent family member(s)	Publication date
US	4822688		18-04-1989	FR	2572417 A	02-05-1986
				DK	495685 A,B,	01-05-1986
				EΡ	0188926 A	30-07-1986
				FI	854172 A,B,	01-05-1986
				JP	4012905 B	06-03-1992
				JP	61113675 A	31-05-1986
				US	4727120 A	23-02-1988
ΕP	0177401	Α	09-04-1986	FR	2570708 A	28-03-1986
				AU	583056 B	20-04-1989
				AU	4791385 A	10-04-1986
				CÁ	1264385 A	09-01-1990
				DK	435585 A	28-03-1986
				ES	547390 A	01-04-1986
				JP	1041165 B	04-09-1989
				JP	1556330 C	23-04-1990
				JP	61089216 A	07-05-1986
				US 	4735992 A	05-04-1988
JP	06128429	A	10-05-1994	NONE		0
EP	0322138	Α	28-06-1989	US	4839422 A	13-06-1989
				US	5010119 A	23-04-1991
				CN	1036391 A	18-10-1989
			•	JP	2001786 A	08-01-1991
				YU	232888 A	31-10-1990